

A petrochemical study of basaltic layering at Henley Harbour, Labrador, using multidimensional scaling

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ABSTRACT

Lack of internal textural evidence for flow tops and small chemical variability among late Proterozoic basalt samples suggest the presence of a single flow at Henley Harbour, Labrador, Canada, despite spectacular, m-scale layering in outcrop. Total chemical variation just slightly exceeds analytical uncertainty for many elements. Multidimensional scaling (MDS) enhances “effective” analytical precision by allowing all data (elements and samples) to be compared simultaneously. Scaling of elements reveals element associations/patterns consistent with igneous controls on distribution, even for alteration-sensitive (Na, K, Sr) elements. MDS also shows that the m-scale layering has a chemical fingerprint. However, to explain chemical variability between layers using mass balance calculations and averaged whole-rock data requires unreasonable (low temperature) phenocryst compositions. Textural features of the m-scale layering record changes in cooling regime between the bottom and center of the flow but chemical variation may reflect changing magma compositions during formation of an inflated flow. However, mass balance calculations suggest that large-scale chemical variation reflects the migration of residual liquid, perhaps as vesicle plumes, within an initially chemically homogeneous flow. Low in the flow (layer 1), metres-long but cm-scale raised and eroded subhorizontal “bands” superficially resemble segregation veins but lack coarse-grained textures. MDS confirms that raised and eroded samples are chemically distinct. Mass balance indicates that banding is not related to phenocryst percentages but may reflect the abundance of a residual liquid resembling the liquid explaining compositional variation between large-scale layers. Alignment of microlites and variations in the quantity of altered residual-liquid glass suggest that banding is related to magma movement during extrusion. It is possible that banding of this type has not been recognized before.

RESUMÉ

L'absence d'attestation texturale interne des sommets d'écoulement et la faible variabilité chimique entre les échantillons de basalte du Protérozoïque tardif laissent supposer la présence d'un écoulement unique à Henley Harbour, Labrador, Canada, malgré la stratification spectaculaire à moyenne échelle de l'affleurement. La variation totale des compositions chimiques dépasse tout juste légèrement l'incertitude analytique de nombreux éléments. L'analyse multidimensionnelle (AMD) améliore la précision analytique « effective » en permettant une comparaison simultanée de toutes les données (éléments et échantillons). L'analyse des éléments révèle des associations/dispositions d'éléments correspondant à des contrôles ignés de la répartition, même dans le cas des éléments sensibles à l'altération (Na, K, Sr). L'AMD révèle par ailleurs que la stratification à moyenne échelle présente une empreinte chimique. L'explication de la variabilité chimique entre les couches au moyen de calculs du bilan pondéral et de moyennes des données sur la roche totale nécessite cependant des compositions de phénocristaux excessives (basse température). Les caractéristiques texturales du fil de stratification à moyenne échelle changent en régime de refroidissement entre le fond et le centre de l'écoulement, mais la variation des compositions chimiques peut correspondre à des compositions magmatiques changeantes pendant la formation d'un écoulement gonflé. Les calculs du bilan pondéral laissent toutefois supposer que la variation des compositions chimiques à grande échelle témoigne de la migration de liquides résiduels, peut-être sous forme de panaches de vacuoles à l'intérieur d'un écoulement initialement homogène du point de vue chimique. Au bas de l'écoulement (première strate), des « bandes » subhorizontales de plusieurs mètres de longueur mais érodées et soulevées de quelques centimètres ressemblent superficiellement à des filons d'exsudation mais sont dépourvues de textures à grain grossier. L'AMD confirme que les échantillons soulevés et érodés sont chimiquement distincts. Le bilan pondéral révèle que le rubanement n'est pas lié au pourcentage de phénocristaux, mais qu'il témoigne de l'abondance d'un liquide résiduel ressemblant au liquide expliquant la variation des compositions entre les strates à grande échelle. L'alignement des microlites et les variations de la quantité de verre soluble résiduel altéré laissent supposer que le rubanement est apparenté à un déplacement magmatique survenu pendant l'extrusion. Il est possible qu'on n'ait pas décelé un rubanement de ce genre auparavant.

INTRODUCTION

The flat-topped basaltic cliffs rising above Henley Harbour, Labrador, have international historical significance as navigational aids (Owen and Greenough 2000). Igneous layering that created the cliffs has been ascribed to extrusion of two or more lava flows representing the surface expression of Late Proterozoic intrusive activity that formed the Long Range dykes (e.g., Strong

and Williams 1972; Strong 1974). Our examination of the cliffs suggested the presence of a single lava flow that could lead to a better understanding of the thermal and chemical processes that produce layering in many individual basaltic flows.

Although apparently common, layering in individual basaltic flows does not tend to be reported by geologists. The subtle or inconspicuous variations in texture give little indication of the potentially large variations in chemical composition between lay-

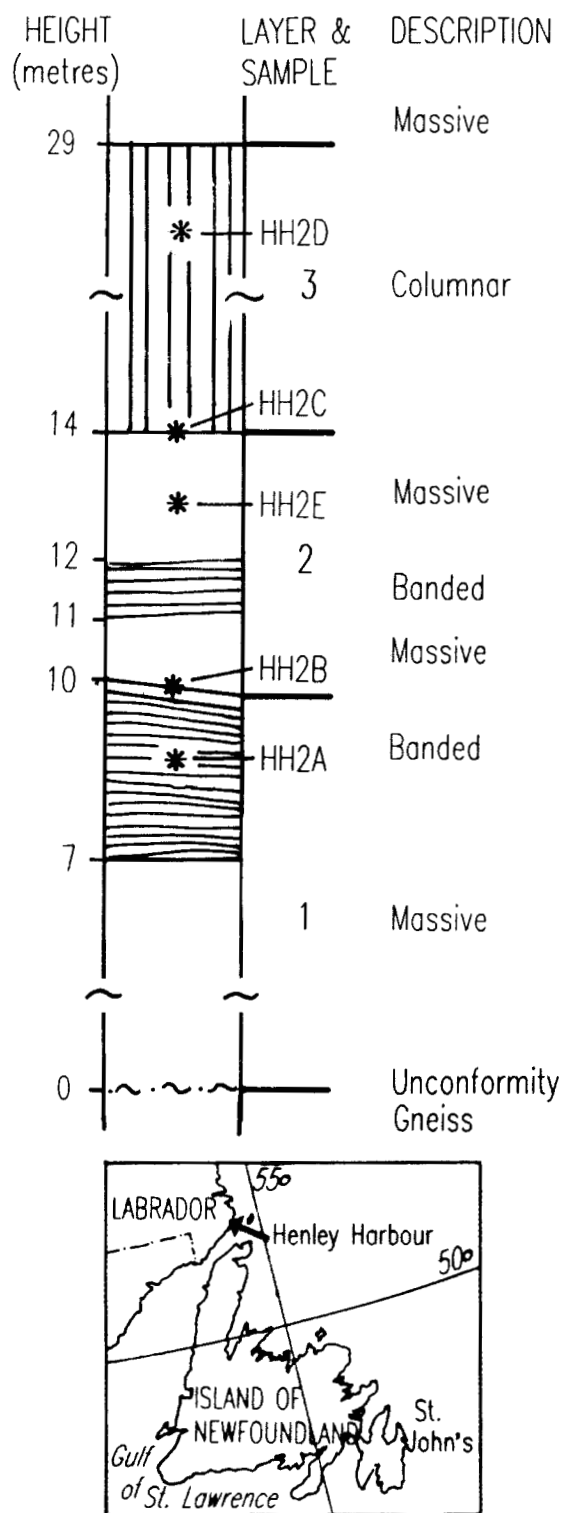


FIG. 1 Stratigraphic section through the Henley Harbour basalt flow showing rock types, layer thicknesses, unit numbers, and sample heights. The section is drawn looking north on the cliff face above Henley Harbour. Note that ~ symbols along the sides of the column denote that, for drafting purposes, portions of the column were omitted. The index map shows the location of Henley Harbour on the Labrador coast.

ers (Goff 1996; Rogan *et al.* 1996; Caroff *et al.* 1997; Greenough *et al.* 1999). Even when reported, layering is likely to be ascribed to multiple extrusions as opposed to cognate processes. An understanding of the physical and chemical processes responsible for layering in lava flows is fundamental due to the information accruing from their rapid cooling regime (Greenough and Dostal 1992; Goff 1996). Processes responsible for layering and chemical variation in lava flows must be processes that act quickly. In contrast, the slow cooling regime of intrusions allows late (slow-acting) processes to overprint and obscure the effects of early, high-temperature, fast-acting processes. However, the study of layering in basaltic lava flows is in its infancy and so it is not surprising that the layering described in this paper has perhaps never before been reported in the literature. Although texturally conspicuous, we use the power of multidimensional scaling to demonstrate that small-scale layering in the basalt at Henley Harbour displays a subtle chemical fingerprint. Similar techniques also show that large-scale layering is probably not reflective of multiple extrusion.

GEOLOGY AND SAMPLING INFORMATION

Plateau basalts of western Newfoundland and southern Labrador, including those at Henley Harbour (Fig. 1), belong to the Late Proterozoic Lighthouse Cove Formation. They are associated spatially and chemically with the Long Range Dyke swarm which cuts gneiss and granitoid rocks of the Long Range inlier in western Newfoundland (Owen 1987) and equivalent Grenvillian rocks in southern Labrador (Strong and Williams 1972; Strong 1974). Both the dykes (Owen *et al.* 1989; Greenough and Owen 1992) and "plateau lavas" have tholeiitic continental flood basalt compositions (Strong and Williams 1972; Strong 1974). Stukas and Reynolds (1974) dated the dykes by $^{40}\text{Ar}/^{39}\text{Ar}$ methods at 605 Ma. A precise U-Pb zircon and baddeleyite date gave 615 ± 2 Ma (Kamo *et al.* 1989). The geochemistry and age of the rocks suggest that they are related to continental rifting and opening of the proto-Atlantic (Iapetus) Ocean (Strong and Williams 1972; Strong 1974; Kamo *et al.* 1989). More recently Puffer (2002) pointed out that the plateau basalts are the most northerly products of a Late Neoproterozoic superplume centered on Sutton Mountain (Quebec) that also yielded basalts as far south as Virginia. Of importance to this study, the rocks at Henley Harbour escaped significant Paleozoic deformation because they lie just west of the "western limit of Appalachian deformation" on the tectonolithofacies map of Williams (1979).

Fig. 1 shows the stratigraphy of the accessible portion of the flow on the north shore of Henley Harbour where our samples were collected. The flow rests unconformably on gneissic rocks. Large-scale (m-scale) layering is apparent from hundreds of metres away, with at least three distinct layers defined by major subhorizontal joints, and traceable across the entire > 100 m wide cliff face (Fig. 2a). Fine-grained, massive, basalt occurs for ~ 7 m above the unconformity. The upper three metres of this ~ 10 m thick (but variable) lower unit contains basalt with raised (resistant) and indented (eroded) bands laterally traceable for 10's of m. The bands resemble bedding (Fig. 2b). Eroded bands appear

thinner than raised bands in the field (Fig. 2b) but laboratory slabs through samples show that both average about 1.5 cm thick. A major subhorizontal “joint” at 10 m separates the lower unit from a second massive unit (~4–5 m thick; Fig. 2b) which thins to the west and shows cm-scale banding ~1 m above its base that resembles banding at the top of layer 1. Another major subhorizontal joint at ~14 m marks the beginning of massive basalt ~15 m thick with well developed columnar (vertical) jointing. The upper portions of this unit were inaccessible (Fig. 2a). In the field it is not obvious what the jointing “breaks” between layers 1, 2 and 3 represent, but vesicular layers, irregular (brecciated) surfaces, or quenched basalt indicative of flow tops were not observed. Table 1 contains the height of samples collected to determine the origin of banding, and chemical and genetic relationships between major layers.

ANALYTICAL METHODS

Mineral compositions were determined using a JEOL Superprobe 733 with four wavelength dispersive spectrometers and an Oxford Link eXL energy dispersive system at the Department of Earth Sciences, Dalhousie University. All elements were detected using the energy dispersive system with a resolution of 137 eV at 5.9 KeV. Each spectrum was acquired for 40 seconds with an accelerating voltage of 15 keV and a beam current of 15 nA. Probe spot size was approximately one micron. The raw data were reduced using Link's ZAF corrections. Precision and accuracy ($\pm 2\%$) were monitored with geological and synthetic reference materials.

For whole-rock analysis, weathered or altered surfaces were removed from samples prior to crushing and powdering in a tungsten-carbide mill. The major elements were determined by X-ray fluorescence (XRF) on fused glass pellets at Memorial University of Newfoundland. Therefore, the totals in Table 1 reflect volatile-free analyses. Precision and accuracy are $\pm 2\%$ or better for most samples, although glasses for a few samples cracked repeatedly (up to 7 attempts at fusion) for unknown reasons, resulting in low totals. Trace elements determined by XRF on pressed powder pellets following Longerich (1995) were run with geological reference materials indicating precision and accuracy better than $\pm 5\%$ for Cl, S, Cu, Zn, Ni, Cr, V, Ga, and Nb and $\pm 12\%$ for Sc. Lithium, Rb, Sr, Ba, Ta, Hf, Zr, Th, U, Y, the rare earth elements (REE), Mo, and Tl were determined using a HF/HNO₃ digestion procedure on 0.1 g of sample with solutions analyzed by Inductively Coupled Plasma-Mass Spectrometry (ICP-MS; Jenner *et al.* 1990). In-run reference materials indicate precision and accuracy better than 5% for all elements except Li and Tl (7%), U (12%) and Mo (18%).

PETROGRAPHY, MINERALOGY, AND SAMPLE DESCRIPTIONS

The samples, which are all very similar, contain augite (≤ 5 mm, \sim Wo₄₀, En₅₀) and plagioclase (≤ 4.5 mm, \leq An₆₉) phenocrysts and glomerocrysts (≤ 2 mm) together comprising less than

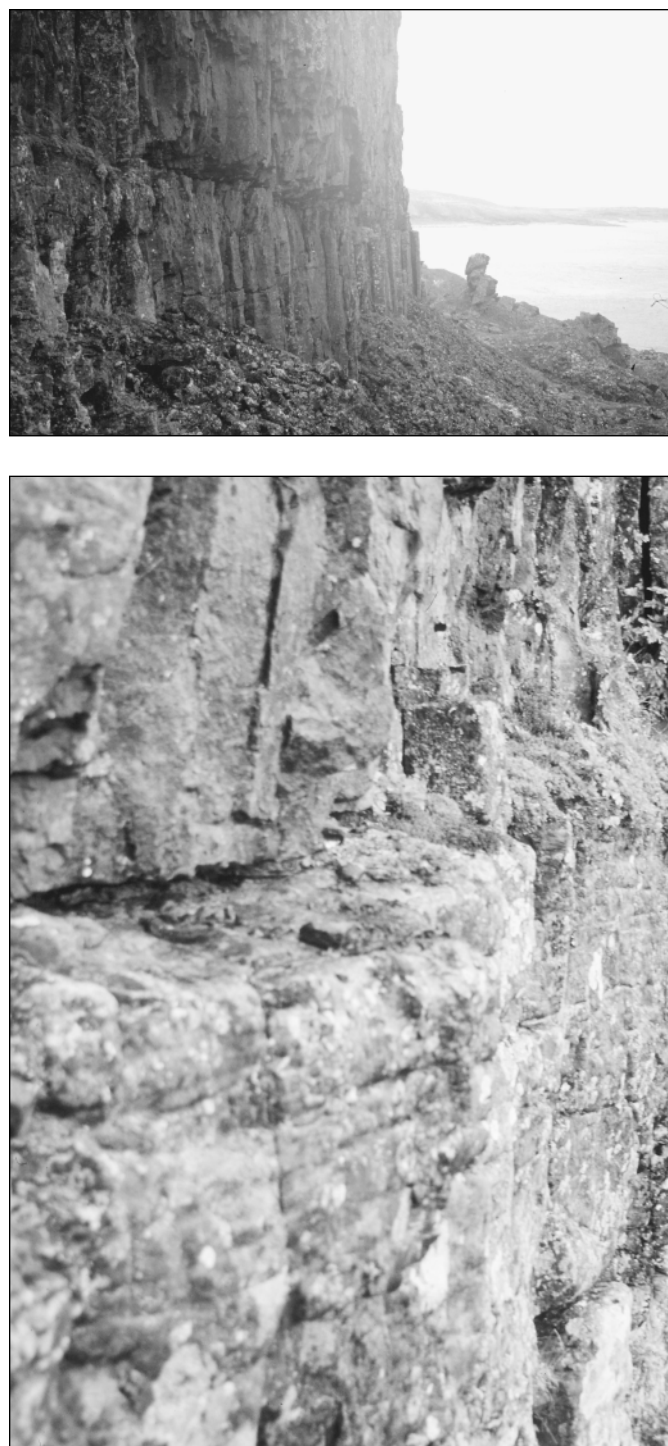


FIG. 2 Photographs of layering in the Henley Harbour flow. a) Columnar jointed basalt of layer 3 is shown below unsampled (inaccessible) massive to columnar jointed basalt higher in the cliff with a sharp boundary between the two in the middle of the photograph. b) Photograph showing the sharp, jointed(?), boundary between banded rocks at the top of layer 1 and the base of more massive basalt of layer 2. Banding in layer 1 (lower portion of photograph) is expressed as sub-parallel, cm-scale, raised and eroded layers.

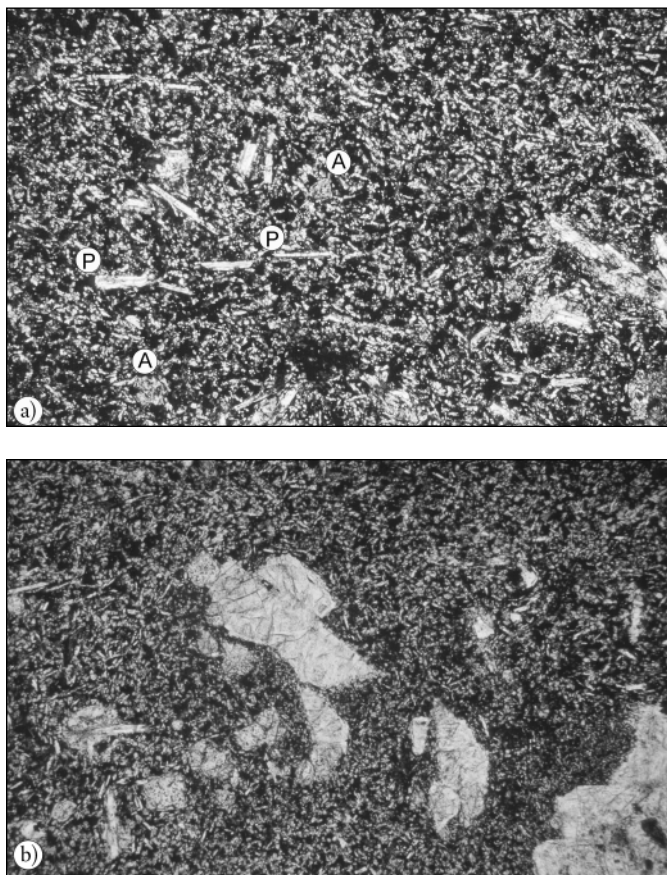


FIG. 3 Photomicrographs of textures in the Henley Harbour flow. a) Photomicrograph showing unaltered elongate plagioclase (P) and augite (A) microphenocrysts in banded sample HH2A. Plagioclase laths display a weak preferred orientation parallel to the upper and lower boundaries of the photograph, which are parallel to the eroded and raised bands in the weathered cliff face. The photograph was taken under crossed polars and is ~5 mm wide. b) Photomicrograph showing a local accumulation of sericitised plagioclase phenocrysts in layer 3 sample HH2D. The picture was taken under crossed polars and measures ~5 mm across.

5% of the rock (Fig. 3a, 3b & 4). Very fine-grained groundmass phases comprise most of each rock and include augite (~0.04 mm), plagioclase (≤ 0.10 mm), and Fe-Ti oxides (≤ 0.1 mm) with modal percentages of 41%, 49%, and 4%, respectively, in least-altered samples. High width/length (w/l) ratios for plagioclase phenocrysts distinguish HH2C, D, and E samples (w/l = 0.3) from HH2A and B samples (w/l = 0.1). With the exception of 0.02 mm wide unaltered rims, plagioclase phenocrysts in HH2C, D, and E tend to be sericitised (Fig. 3b). Both eroded and raised "banded" samples (HH2A) show an alignment of feldspar laths (microphenocrysts) parallel to bands (Fig. 3a). The only difference between raised versus eroded bands in HH2A is that very fine-grained alteration minerals (clay minerals?) are olive green to brown in the former (7 modal %) and forest green in the latter

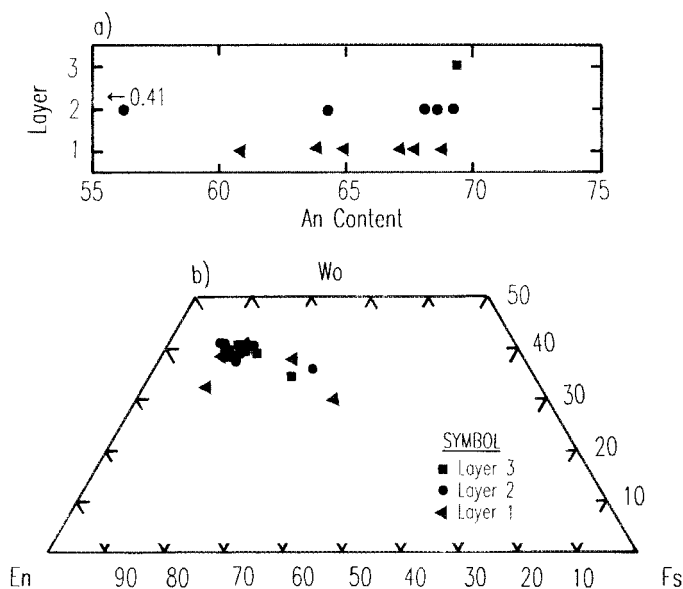


FIG. 4 Plots of plagioclase (a) and augite (b) compositions. An content refers to the atomic percentage of anorthite component in plagioclase. En, Wo, and Fs stand for enstatite, wollastonite and ferrosilite components of augite, respectively.

bands (16%). The olive green mineral represents the dominant alteration mineral ($\leq 7\%$) in HH2C, D, and E.

WHOLE-ROCK DATA

Loss on ignition values (LOI, Table 1) range from ~1.1 to 2.6 wt. % and show a positive correlation with a number of elements (e.g., MgO, Na₂O, Zr; Fig. 5). CIPW norms contain hypersthene (rocks are subalkaline; assuming Fe₂O₃/FeO = 0.18; Irvine and Baragar 1971) which is consistent with tholeiitic Nb/Y ratios of ~0.3 and TiO₂ concentrations (~1.3 wt. %) within the range of continental tholeiite (Pearce and Cann 1973). Henley Harbour element concentrations plot at the high Mg# (= Mg/(Mg + 0.9*Fe^{total}) atomic) end of trend lines for the Long Range dykes on variation diagrams, and element ratios such as TiO₂/Zr are very similar in both groups of rocks (Fig. 6). Total chemical variation within the flow is small in comparison with variation between the Long Range Dykes (Fig. 6). Trend lines are apparent for some elements on variation diagrams (e.g., Y), but for other elements they are only poorly defined (e.g., TiO₂; Fig. 7). Further, samples that were obviously distinguishable in the field (e.g., Layer 1 Eroded and Raised samples) do not appear chemically distinguishable (Fig. 7).

STATISTICAL METHODS

As elaborated on in the Discussion, the possibility of alteration effects on the geochemistry, and the limited range of chemical

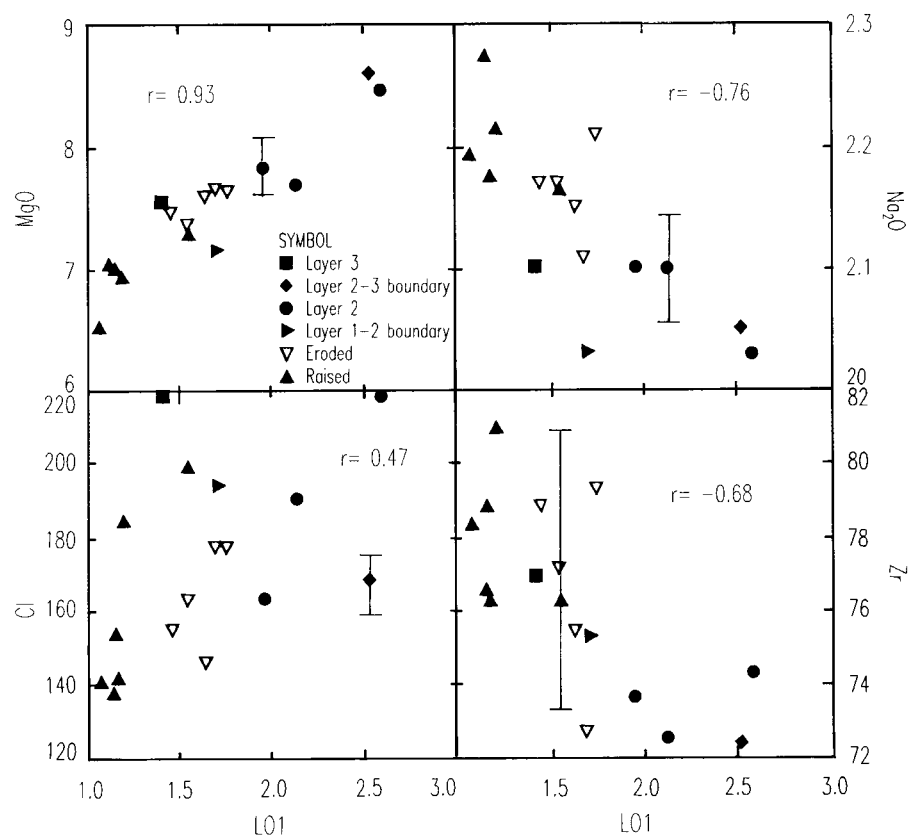


FIG. 5 Plot of LOI (loss on ignition) versus selected major element oxides (MgO and Na₂O in wt. %) and trace elements (Zr and Cl in ppm). Pearson correlation coefficients, r , are shown for each plot along with the + and - 1 σ error bars on the data.

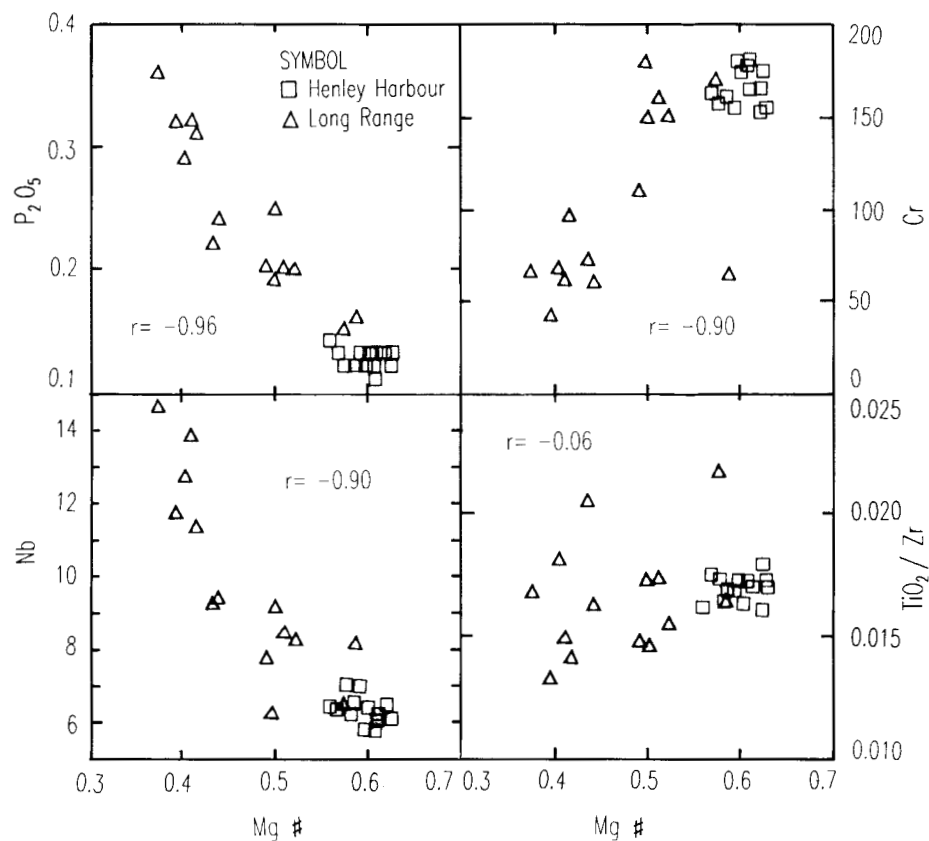


FIG. 6 Plot of Mg# values ($\text{Mg\#} = \text{Mg}/(\text{Mg} + 0.9 \cdot \text{Fe}^{\text{total}})$ atomic) versus P₂O₅ (oxide wt. %), Cr and Nb (ppm) concentrations and TiO₂/Zr ratios (oxide wt. %/ppm) for Henley Harbour samples and the Long Range Dykes. Pearson correlation coefficients (r values) are shown for each graph. Long Range Dyke data are from Greenough and Owen (1992).

Table 1: Major element and trace element analyses of Henley Harbour basalt.

	HH2AR1	HH2AE1	HH2AR2	HH2AR3	HH2AE3	HH2AR4	HH2AE4	HH2AR5	HH2AE5	HH2AR6
Layer ¹	1	1	1	1	1	1	1	1	1	1
Height ²	8.73	8.75	8.77	8.84	8.87	8.89	8.91	8.93	8.96	8.98
Wt %										
SiO ₂	52.58	51.16	51.34		48.95	50.01	49.95	47.72	50.26	50.38
TiO ₂	1.30	1.35	1.37		1.25	1.32	1.31	1.27	1.30	1.34
Al ₂ O ₃	14.71	14.63	14.39		13.90	14.18	14.39	13.70	14.48	14.30
Fe ₂ O ₃	9.16	9.93	9.73		10.73	11.18	11.44	11.23	11.39	11.61
MnO	0.18	0.16	0.19		0.18	0.19	0.19	0.18	0.19	0.19
MgO	6.93	7.64	7.29		7.67	6.99	7.46	6.52	7.60	7.02
CaO	11.25	11.29	11.14		11.04	11.33	11.24	10.99	11.27	11.61
Na ₂ O	2.21	2.21	2.16		2.11	2.17	2.17	2.19	2.15	2.27
K ₂ O	0.75	0.64	0.72		0.49	0.63	0.63	0.65	0.62	0.64
P ₂ O ₅	0.12	0.13	0.13		0.11	0.12	0.13	0.14	0.13	0.13
Total	99.19	99.14	98.46		96.43	98.12	98.91	94.59	99.39	99.49
LOI	1.20	1.75	1.55	1.16	1.69	1.17	1.45	1.07	1.64	1.14
Mg#	0.63	0.63	0.62		0.61	0.58	0.60	0.56	0.60	0.57
ppm.										
Cl	184	178	199	153	178	141	155	140	147	137
S	219	187	199	240	231	106	108	124	106	113
Mo	0.320	0.407	0.282	0.291	0.353	0.328	0.362	0.380	0.447	0.332
Cu	44.7	52.5	44.3	973	718	143	148	136	145	130
Zn	46.0	53.3	53.7	58.5	72.9	62.8	68.5	59.7	69.1	57.7
Tl	0.064	0.064	0.066	0.146	0.117	0.100	0.109	0.096	0.111	0.107
Ni	58.3	75.8	57.2	69.1	70.9	68.6	68.6	62.4	71.2	66.3
Cr	153	155	166	165	165	158	154	154	153	163
Sc	49	44	47	47	43	42	40	38	44	43
V	233	266	268	254	256	257	261	251	264	260
Ga	19.1	17.2	18.4	17.5	17.2	19.1	18.7	18.6	18.8	18.5
Li	16.4	14.7	15.2	13.6	20.0	11.2	14.9	10.4	15.2	11.0
Rb	13.9	9.4	11.7	9.7	7.6	9.9	10.1	9.4	9.1	9.4
Sr	246	239	222	237	230	239	238	239	229	238
Ba	285	249	304	242	248	245	256	254	251	254
Nb	6.09	6.04	6.41	6.41	5.90	6.94	7.23	6.37	6.61	6.31
Ta	1.01	1.53	1.37	1.18	1.06	0.95	0.91	1.03	0.84	0.98
Zr	80.9	79.4	76.3	78.8	72.7	76.2	78.9	78.3	75.5	76.5
Hf	2.17	2.40	2.34	2.23	2.21	2.29	2.47	2.43	2.21	2.25
Th	0.606	0.659	0.692	0.657	0.624	0.668	0.661	0.683	0.671	0.629
U	0.141	0.148	0.182	0.147	0.152	0.173	0.140	0.158	0.164	0.162
Y	21.9	22.6	21.6	21.5	20.8	21.6	22.1	21.5	21.6	21.4
La	8.36	8.16	8.75	8.44	7.61	8.33	8.33	8.64	7.98	8.45
Ce	19.1	19.1	19.7	19.5	18.5	19.6	19.6	20.1	19.4	19.7
Pr	2.57	2.67	2.71	2.65	2.64	2.72	2.72	2.83	2.72	2.74
Nd	11.8	12.2	12.5	12.3	11.9	12.5	12.9	12.4	12.3	12.3
Sm	3.18	3.27	3.34	3.30	3.07	3.30	3.33	3.36	3.31	3.24
Eu	1.14	1.26	1.16	1.15	1.10	1.21	1.18	1.17	1.15	1.12
Gd	3.67	3.92	3.86	3.67	3.88	3.94	3.92	3.93	3.85	3.74
Tb	0.582	0.640	0.623	0.664	0.609	0.627	0.665	0.641	0.631	0.656
Dy	3.83	4.18	4.17	4.13	3.94	4.13	4.17	4.23	4.12	4.03
Ho	0.827	0.923	0.881	0.885	0.83	0.883	0.87	0.857	0.865	0.836
Er	2.23	2.52	2.52	2.5	2.44	2.5	2.54	2.6	2.47	2.38
Tm	0.324	0.363	0.347	0.349	0.341	0.351	0.363	0.369	0.332	0.342
Yb	2.1	2.28	2.25	2.24	2.18	2.28	2.32	2.23	2.21	2.21
Lu	0.303	0.351	0.346	0.334	0.347	0.334	0.344	0.329	0.334	0.327

Notes: Major element oxides in wt. % volatile free with total Fe as Fe₂O₃ except averages recalculated to 100% volatile-free.

LOI = loss on ignition. Trace elements in ppm. Mg# = $\text{Mg}/(\text{Mg}+0.9\text{Fe}^{\text{total}})$ atomic.

¹ Numbers give the layer each sample came from. T samples came from the very top of the layer.

² Height = height above the base of the flow in metres.

Table 1. (contd.)

Layer ¹	HH2AE6	HH2B	HH2ET	HH2EM	HH2EB	HH2C	HH2D	Average	Average	Average
Height ²	1	1T	2	2	2	2T	3	1 Raised	1 Eroded	2
	8.99	10	13.87	13.72	13.6	13.97	16.71	8.86	8.9	13.79
Wt %										
SiO ₂	49.13	48.56	48.12	49.74	49.53	49.02	49.72	51.45	50.77	49.61
TiO ₂	1.30	1.24	1.27	1.27	1.25	1.25	1.26	1.35	1.32	1.27
Al ₂ O ₃	14.12	14.17	15.11	15.00	14.97	14.97	14.73	14.55	14.57	15.17
Fe ₂ O ₃	11.29	11.41	11.73	11.07	11.30	11.29	10.99	10.80	11.15	11.46
MnO	0.19	0.19	0.24	0.21	0.21	0.20	0.18	0.19	0.19	0.22
MgO	7.36	7.17	8.48	7.85	7.71	8.62	7.58	7.09	7.68	8.25
CaO	11.18	11.17	11.18	11.35	11.31	11.21	11.79	11.50	11.40	11.38
Na ₂ O	2.17	2.03	2.03	2.10	2.10	2.05	2.10	2.25	2.20	2.09
K ₂ O	0.60	0.39	0.39	0.42	0.45	0.44	0.56	0.69	0.61	0.43
P ₂ O ₅	0.12	0.12	0.13	0.12	0.12	0.12	0.13	0.13	0.13	0.12
Total	97.46	96.45	98.68	99.13	98.95	99.17	99.04	100	100	100
LOI	1.54	1.71	2.59	1.96	2.13	2.53	1.41			
Mg#	0.59	0.59	0.61	0.61	0.60	0.63	0.60	0.59	0.61	0.61
ppm.										
Cl	163	194	219	164	191	169	219	159	164	186
S	110	72	56	99	76	56	230	167	148	72
Mo	0.320	0.298	0.231	0.259	0.239	0.290	0.220	0.322	0.378	0.255
Cu	150	129	64.2	170	115	52.3	123	245	243	100
Zn	63.9	70.0	71.7	59.8	53.6	66.5	72.9	56.4	65.5	62.9
Tl	0.120	0.044	0.043	0.048	0.043	0.054	0.088	0.096	0.104	0.047
Ni	69.4	67.5	74.4	64.0	62.4	70.6	74.4	63.6	71.2	67.8
Cr	161	160	182	180	181	175	175	160	158	180
Sc	37	33	41	46	41	43	34	44	42	43
V	258	271	276	275	274	283	279	254	261	277
Ga	20.6	18.4	16.6	16.8	15.9	17.3	17.8	18.5	18.5	16.6
Li	15.5	16.0	25.5	21.6	21.1	26.0	19.0	13.0	16.1	23.6
Rb	10.2	9.5	9.9	10.3	11.2	10.8	15.4	10.7	9.3	10.6
Sr	237	235	229	241	232	230	246	237	235	233
Ba	257	176	212	209	202	232	216	264	252	214
Nb	6.45	6.15	6.15	5.73	5.76	6.03	6.30	6.42	6.45	5.92
Ta	1.13	1.06	0.95	0.68	0.96	0.89	0.81	1.09	1.09	0.87
Zr	77.2	75.3	74.3	73.7	72.6	72.4	77.0	77.8	76.7	73.2
Hf	2.25	2.22	2.22	2.20	2.17	2.12	2.25	2.28	2.31	2.18
Th	0.618	0.634	0.597	0.632	0.599	0.599	0.647	0.656	0.647	0.607
U	0.158	0.192	0.151	0.117	0.125	0.145	0.163	0.160	0.150	0.130
Y	21.2	20.7	21.1	20.9	20.1	20.5	21.5	21.6	21.7	20.6
La	7.85	8.63	8.37	8.23	8.33	7.83	7.98	8.50	7.99	8.19
Ce	19.2	20.4	19.5	19.7	19.5	18.4	20.6	19.6	19.2	19.3
Pr	2.66	2.76	2.65	2.68	2.58	2.58	2.86	2.70	2.68	2.62
Nd	12.4	12.3	12.3	12.1	12.0	11.8	13.1	12.3	12.3	12.0
Sm	3.28	3.31	3.31	3.29	3.12	3.21	3.35	3.29	3.25	3.23
Eu	1.17	1.21	1.15	1.17	1.17	1.24	1.27	1.16	1.17	1.18
Gd	3.80	3.75	3.73	3.78	3.69	3.72	3.72	3.80	3.87	3.73
Tb	0.613	0.667	0.624	0.619	0.590	0.616	0.652	0.632	0.632	0.612
Dy	4.02	4.04	3.95	3.94	3.96	4.00	4.10	4.09	4.09	3.96
Ho	0.856	0.846	0.842	0.821	0.801	0.812	0.87	0.862	0.869	0.819
Er	2.47	2.39	2.47	2.41	2.34	2.31	2.46	2.46	2.49	2.38
Tm	0.342	0.342	0.345	0.33	0.334	0.337	0.34	0.347	0.348	0.336
Yb	2.19	2.18	2.15	2.19	2.15	2.21	2.31	2.22	2.24	2.18
Lu	0.362	0.339	0.341	0.330	0.308	0.332	0.325	0.329	0.348	0.328

variability among samples complicates the interpretation of the data set in terms of igneous processes. One potential way to overcome these problems is to look for data patterns using all analytical data simultaneously. Multidimensional scaling (MDS) is an exploratory statistical technique with an ability to represent measurements of similarity (or dissimilarity) between "objects" in a small number of dimensions (Borg and Groenen 1997, pp. 1–14). It is a method well known to social scientists (*ibid.*) but we have discovered that it is also remarkably efficient at uncovering patterns in large geochemical data sets. For example, it can be used to illustrate overall geochemical similarities and differences between chemical elements in a two-dimensional space using all data (Table 1) from all samples (Fig. 8). Elements that behave in a similar way (i.e. when one element increases in a sample, an element behaving similarly would also increase) plot close together on the MDS diagram, whereas chemically dissimilar elements fall far apart (Fig. 8). Data processing used SYSTAT statistical software (Wilkinson *et al.* 1992, pp. 109–131). Distances between elements were determined from a matrix of Pearson correlation coefficients calculated for all possible element pairs (e.g., Rb:Zr) using standardized data (all elements were put on the same scale using z-scoring). Manually added fields in Fig. 8 bear no calculated statistical significance but emphasize that elements with known geochemical similarities occur close to one another. For example, the REE plot together and close to a second group that includes most high-field-strength elements (Nb, Zr etc.; Fig. 8). Other calculation and processing details appear in the caption for Fig. 8.

Multidimensional scaling can also be used to examine relationships between samples (Fig. 9). As for Fig. 8, all elements in the data set were put on the same scale using z-scoring, but in this case the data matrix was then inverted and concentrations in each sample compared to the concentrations in other samples creating a sample-versus-sample matrix of Pearson correlation coefficients. Details of the calculations appear in the caption for Fig. 9. Manually added fields show that Layer 1 samples are apparently chemically distinct from layer 2 samples. The layer 3 sample (Fig. 1) is intermediate between layer 1 and layer 2 samples but most similar to layer 1. Data were also processed in other ways to confirm that these conclusions are not dependent on the type of similarity/dissimilarity measure utilized (e.g., Euclidean distances versus the Pearson correlation coefficients used) or the MDS calculation procedure (e.g., Kruskal versus Guttman loss function; monotonic versus linear distance function). Large-scale physical/textural layering in the flow clearly has a chemical signature (Fig. 9).

Fig. 10 uses multidimensional scaling to examine relationships between a subset of samples, the "banded" samples from layer 1. To ensure that uncovered patterns are not due to alteration effects, a subset of elements known to be "immobile" during low-grade metamorphism (Ti, P, Ni, Cr, Sc, V, Ga, Nb, Ta, Zr, Hf, Y, and selected REE = La, Nd, Dy, Yb) was used in processing. In other respects the calculation procedures were the same as for Fig. 9. Manually added fields show that eroded samples are distinct from raised samples. Further, lines connecting contiguous

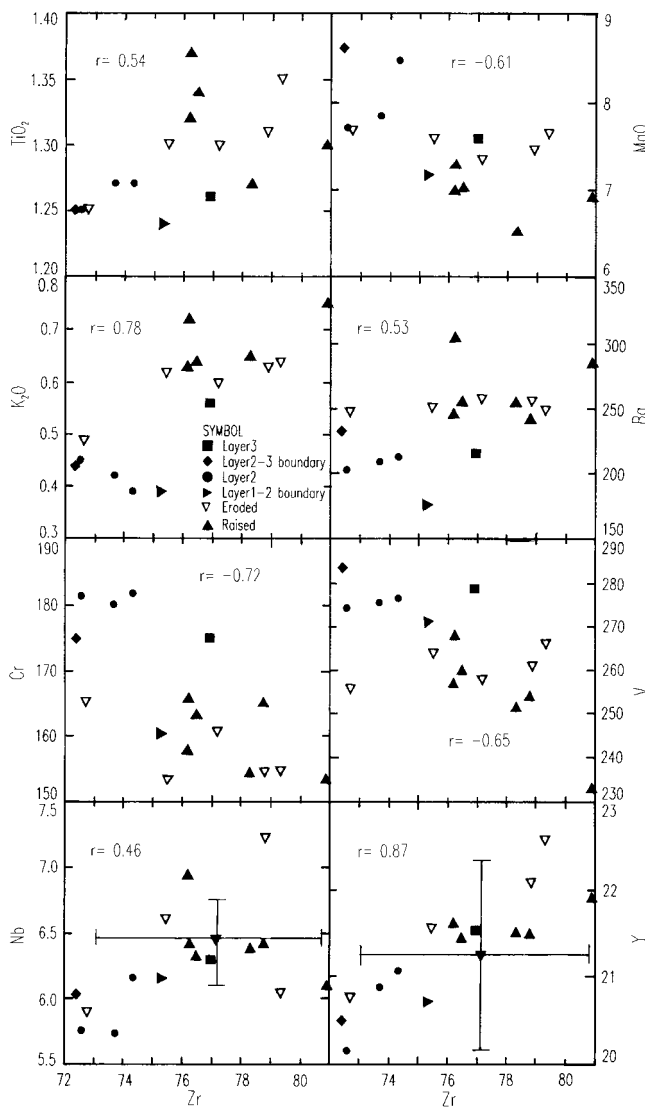


Fig. 7. Plot of Zr concentrations (ppm) versus selected major element oxides (wt. %) and trace elements (ppm) with a variety of geochemical characteristics. Data on the graphs have Pearson correlation coefficients (r values) as shown. The two lower graphs provide examples of the $+1\sigma$ and -1σ error bars on data. See text for discussion.

raised and eroded samples demonstrate that there is "pairing" between spatially associated layers (Fig. 10). Thus Dimension 1 in the diagram separates samples according to whether they are raised or lowered whereas Dimension 2 gives a measure of how close samples were to one another in the stratigraphy. Note that the use of all elements instead of just "immobile" elements yields an analogous MDS diagram.

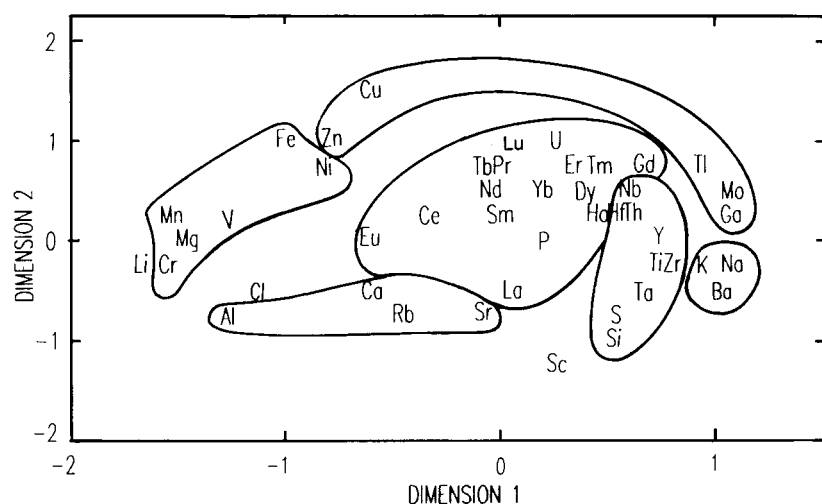


FIG. 8 Multidimensional scaling (MDS) diagram which places elements that behaved similarly in the Henley Harbour data set close together and dissimilar elements far apart. Fields around elements were added manually to illustrate the tendency for elements to group according to geochemical character as discussed in the text. The diagram was created by z-scoring the data and calculating a similarity/dissimilarity matrix (pairwise deletion) from Pearson correlation coefficients. Scaling was done in 2 dimensions with a Kruskal loss function. A monotonic function of distances was fit to the input data matrix. The solution yielded a straight-line Shepard diagram and required 36 iterations with stress diminishing smoothly from 0.256 to 0.204.

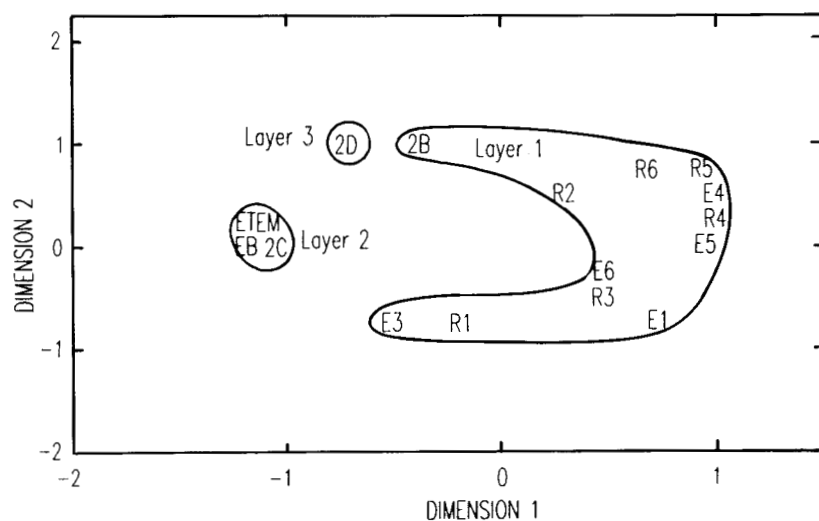


FIG. 9 MDS diagram comparing the overall chemical similarity between samples from layers 1, 2 and 3. Fields shown on the diagram were added manually and demonstrate that samples from each physically identified layer are chemically most similar. Samples are identified using concatenations of the sample numbers in Table 1. Processing involved z-scoring to put all elements on the same scale and then calculating a sample versus sample similarity/dissimilarity matrix (pairwise deletion) composed of Pearson correlation coefficients. Monotonic scaling was done in 2 dimensions with a Kruskal loss function and yielded a straight-line Shepard diagram with stress diminishing smoothly from 0.233 to 0.187 over 17 iterations.

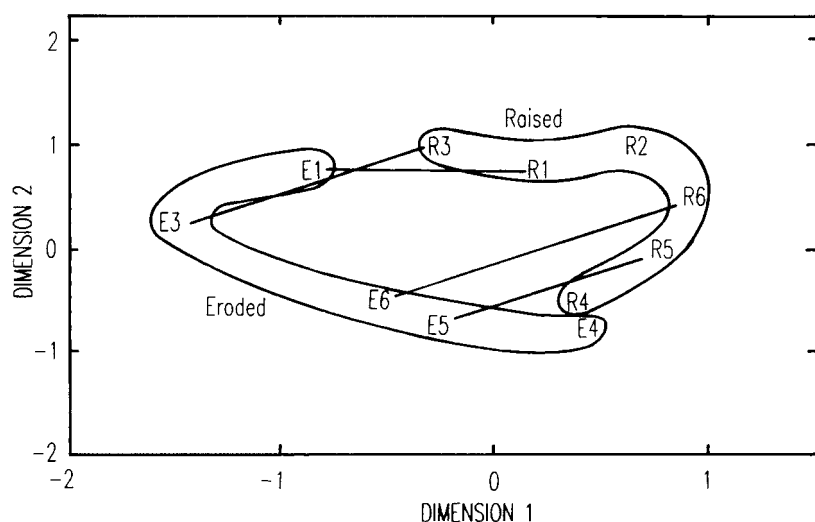


FIG. 10 MDS diagram illustrating the overall chemical relationships between eroded and raised samples from layer 1. Fields on the diagram surround the two sample types and were drawn manually. Only relatively "immobile" elements (Ti, P, Ni, Cr, Sc, V, Ga, Nb, Ta, Zr, Hf, Y, and selected REE = La, Nd, Dy, Yb) were used in the calculations to demonstrate that the layering is probably not related to alteration processes. Sample numbers are abbreviated and should have HH2A in front of them. Lines connect samples from adjacent raised and eroded layers. Processing involved z-scoring to put all elements on the same scale followed by calculation of a sample versus sample similarity/dissimilarity matrix (pairwise deletion) composed of Pearson correlation coefficients. Monotonic scaling in 2 dimensions used a Kruskal loss function and yielded a straight-line Shepard diagram with stress diminishing smoothly from 0.165 to 0.132 over 13 iterations.

DISCUSSION

Alteration

High percentages of igneous plagioclase and augite in all samples attest to excellent preservation for Late Proterozoic rocks. Similarly, LOI values are low (Table 1). However, MgO and LOI show the strongest correlation of any analyses in the data set with a correlation coefficient (r) of 0.93. This correlation may reflect the greater probability of chlorite formation (and water addition) in high MgO samples. This process also explains other correlations with LOI. For example, the moderately strong Zr-LOI correlation ($r = -0.68$) is inconsistent with the randomizing effect of alteration processes and the immobile nature of Zr. In fact, most correlations between elements (Fig. 7), though weak, suggest igneous controls on data patterns. Nevertheless, it is not possible to prove using this approach that secondary processes have not affected some element concentrations (e.g., Cl) in the basalt.

The problem with interpreting chemical differences among these samples is that total concentration variations for many elements approach the limits for analytical uncertainty as shown by the + and -1σ error bars on Fig. 5 and Fig. 7. For example, the total range in Zr concentrations is 72.6 to 80.9 ppm with a median value of 76.8 (Fig. 7; Table 1). If analytical uncertainty is as poor as $\pm 5\%$, the 1σ error bars around the median represent 93% of the total range of values observed. To help interpret chemical variation within the flow in terms of either primary or secondary processes some method of filtering out analytical uncertainty is required. The discussion below suggests that multidimensional scaling may be extremely useful.

Fig. 8 shows relationships between all elements in all samples using multidimensional scaling techniques. Manually added fields illustrate that the REE plot together and close to a second group that includes most high-field-strength elements (Nb, Zr etc.; Fig. 8). Elements associated with the ferromagnesian minerals also occur together. Those elements controlled by feldspars form two separate groups. The association of Ca, Sr and Al makes sense because high Ca concentrations in plagioclase are accompanied by elevated Al concentrations effecting charge balance. Strontium partitioning coefficients tend to be greater than unity in plagioclase (Philpotts, J.A., and Schnetzler 1970). These elements plot close to the relatively immobile element Eu, which in the 2+ state can partition into plagioclase (Drake and Weill 1975). Sodium (K and Ba) are separated from the first feldspar-related group because as Ca goes up Na goes down in plagioclase. Similarly, mass balance dictates that samples slightly enriched in Ca-rich plagioclase phenocrysts would be depleted in Na, K, and Ba due to displacement of residual liquid by the phenocrysts. Samples slightly enriched in residual liquid would be enriched in these elements which tend to be incorporated in late-forming feldspars. Although Rb plots with elements associated with feldspars, it would be more reasonable to see it with the alkali metals, Na and K. The anomalous locations of Li and Sc could reflect lower analytical precision for these elements than most others. In summary, with the exception of a few elements (Li, Rb, Cl, S, and Sc) most chemical data in Fig. 8 associate as would be predicted for the minerals in unaltered

basaltic rocks. Natural chemical systems that have been disturbed by human activities (Mallory-Greenough *et al.* 1998) or the randomizing (disequilibrium) effects of partial alteration would not be expected to show such regular behaviour of elements on the MDS diagram.

Relationships between layering and chemical variation

Henley Harbour geochemical data plot at the high Mg# end of variation diagram trend lines for the Long Range Dykes (Fig. 6). This observation is consistent with the conclusions of Strong and Williams (1972) and Strong (1974) that the basalts are extrusive equivalents of the dykes. They are primitive (higher Mg# and Cr as well as lower incompatible element concentrations) compared to the dykes and exhibit an extremely limited range of compositions. More chemical variation occurs across an individual Long Range Dyke than in all basalt samples from Henley Harbour. For example TiO₂ ranges from 1.2 to 1.4 wt. % at Henley Harbour, but concentrations in dyke 622 vary between 2.8 and 3.5 wt. % (Table 1 in Greenough and Owen 1992). This observation suggests, but certainly does not prove, that a single flow exists at Henley Harbour.

The chemical argument for a single flow is supported by field information. Although the basalt is clearly "layered" (Fig. 1), there is no evidence for variations in vesicle percentages, grain size differences, oxidation, or surface irregularities between units that might indicate a flow top in the studied portion of the flow (Fig. 1).

Are there chemical variations at Henley Harbour that reflect the layering observed in outcrop? Multidimensional scaling confirms that the subtle chemical variations portrayed in Fig. 7 are associated with overall chemical variations within the flow (Fig. 9) and that the large-scale (m-scale) layers observed in the field have distinct chemical signatures. Chemical variations among eroded and raised bands are even more subtle than between the large-scale layers (Fig. 7) and it is not clear from Table 1 whether they have chemical significance or simply textural significance. The band-parallel fabric exhibited by feldspar microphenocrysts and the observation that eroded and raised bands are traceable for metres across the outcrop face strongly suggests that these features are not related to alteration processes. The MDS plot for the fine-scale banded samples alone demonstrates that raised layers are distinct from eroded layers (Fig. 10). The plot even reproduces the hand sample-scale stratigraphy shown by raised and eroded samples. Remarkably, these chemical - textural - stratigraphic relationships in banded samples are not apparent from the variation diagrams (Fig. 7), emphasizing the power of MDS to identify subtle similarities and differences between chemically similar samples.

The origin and significance of banding in layer one

The fine-scale banding is expressed on weathered outcrop surfaces as cm-scale raised and eroded layers that superficially

resemble some interlayered basalt- (eroded layers) segregation veins (raised) reported from basaltic flows (Anderson *et al.* 1984; Helz 1980, 1987; Greenough and Dostal 1992; Puffer and Horter 1993; Goff 1996; Rogan *et al.* 1996; Greenough *et al.* 1999; Puffer and Volkert 2001). However, Henley Harbour “bands” are probably not segregation veins because they are too fine-grained. Other features such as their stratigraphic regularity, location in the flow (lower portion), and lack of vesicles, although not individually inconsistent with a segregation vein model, together suggest some other origin.

The banded rocks show an alignment of plagioclase microphenocrysts, possibly resembling compacted plagioclase chain networks associated with melt expulsion in the colonnade of thick basaltic lava flows (Philpotts *et al.* 1996, 1999; Philpotts and Dickson 2000; Puffer and Volkert 2001). However, compaction has not been found associated with cm-scale, regular banding and the Henley Harbour banded rocks are very fine-grained with small percentages of high-aspect-ratio plagioclase laths (Fig. 3a), features that are texturally distinct from what has been described for collapsed and compacted colonnade basalts.

Banding and plagioclase alignment macroscopically and microscopically resemble features observed in multiply injected basaltic dykes. Thin sections show that the Henley Harbour banding is expressed as differences in the quantity and composition-related pleochroism of extremely fine-grained interstitial secondary clay(?) minerals, apparently replacing glass. The plagioclase microlites are aligned parallel to the bands. Greenough and Hodych (1990) reported repeating bands of quenched glass and very fine-grained basalt near the contacts of a Mesozoic dyke in Atlantic Canada. These features, which are on a similar scale to the banding at Henley Harbour, were related to magma pulsing, or multiple magma injection through the basaltic dyke. Feldspar alignment in these rocks made it possible to infer the direction of magma movement through the dyke because numerous observations and experiments have related the long axis of phenocrysts in fine-grained (non-cumulus) mafic rocks with flow directions (e.g., Clark 1952; Bhattacharji 1967; Ross 1986). The textural features reported by Greenough and Hodych (1990) are thus very similar to those at Henley Harbour and suggest that variations in the percentage of interstitial glass and alignment of plagioclase microlites reflect pulsing of magma through a conduit or lava flow.

The multidimensional scaling diagram (Fig. 10) and Table 1 averages suggest that, along with textural differences between raised and eroded bands, there are chemical differences. If feldspar alignment resulted from episodic magma movement along the floor of the flow, it is possible that compositional variations reflect repeated (pulse-related), compaction with Bernoulli-effect extraction of interstitial liquid out of eroded (residue) bands, and into overlying, moving magma (Fig. 11). In other words, as magma pulsed through the flow, it repeatedly and regularly caused residual liquid extraction, from an advancing, semi-solidified crystallization front at the base of the flow. It is unlikely that variations in the composition between raised and eroded bands can simply be ascribed to variations in phenocryst percentages for two reasons. First, there is no petrographic evidence that raised and eroded samples have significantly different phenocryst con-

tents. Secondly, mass balance calculations (Bryan *et al.* 1969) indicate that phenocryst accumulation may not be able to account for differences in composition between raised and eroded bands. Calculations used the augite and plagioclase analyses portrayed in Fig. 4. The average of eroded samples HH2AE1, HH2AE4, HH2AE5, and HH2AE6 represented “parent” and the average of the associated raised samples (HH2AE1, HH2AE4, HH2AE5, and HH2AE6) acted as “daughter”. The calculations produced low r^2 values (0.2; sum of squared residuals) only if the most Fe-rich augite and Na-rich plagioclase compositions available were used and the latter almost certainly do not reflect the average composition of plagioclase phenocrysts in the samples. If a residual liquid was extracted it must represent liquid extracted from around groundmass phases. Mass balance calculations in the following section provide further support for the hypothesis that variations in composition between raised and eroded bands are due to movement of evolved interstitial residual liquid.

Banding similar to that at Henley Harbour has apparently not been previously reported. It has been argued here that the chemistry of the bands reflects primary processes, in particular small differences in the percentage of residual interstitial liquid (glass). However, without the secondary alteration which facilitated differential weathering, it is not obvious that this type of banding would be visible for sampling in totally unaltered and unweathered rocks. In this respect it may resemble segregation vein layering in a dacite lava flow at Kelowna, British Columbia, Canada, that would be very difficult to detect in a fresh flow due to minuscule chemical differences between the segregation veins and associated dacite (Greenough and Owen 1998).

The origin and significance of large-scale layering

The lack of samples from the inaccessible portions of the Henley Harbour flow makes it difficult to mass balance chemical compositions and create a quantitative model for element distributions and large scale layering such as done, for example, with Kilauea Iki (Helz *et al.* 1989). Nevertheless, models can be semi-quantitatively evaluated. In terms of MgO content, layer 1 is slightly more evolved than layer 3, and layer 2 is the most primitive (Table 1). It is possible that the base of the flow represents founded crust. Helz *et al.* (1989) reported that founded crust at the base of Kilauea Iki Lava Lake has highly variable composition. However, the founded crust appears more primitive (higher MgO) than the average composition of the lava lake because of high olivine phenocryst abundances. Although it is possible that layer 1 represents founded crust that was less primitive than layer 2, it has already been pointed out that there is no evidence for variations in vesicle percentages, grain size differences, oxidation, or surface irregularities between units that might indicate a flow top and founded crust.

Chemical variation between layers 1 and 2 could also reflect changes in the composition of magma during eruption. For example, changes in magma composition have been documented for the 1959 eruption of Kilauea Iki Lava Lake (Wright 1973) although pre-eruptive heterogeneity was largely eliminated during eruption

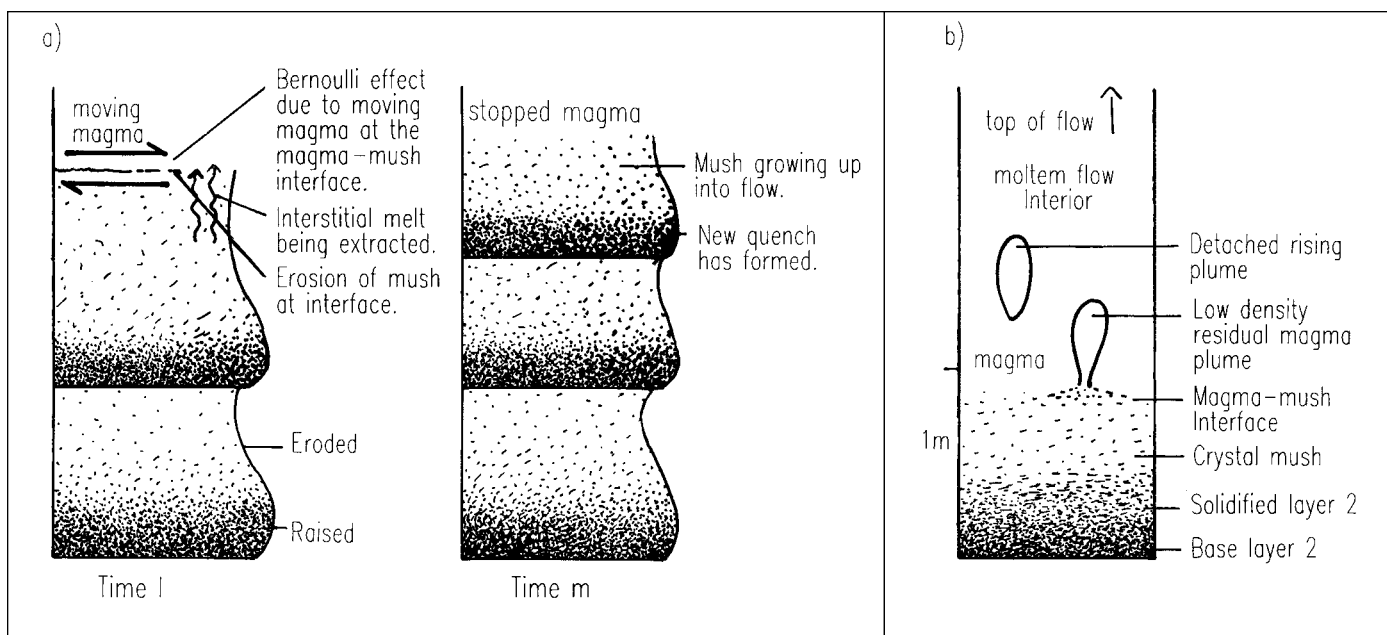


Fig. 11. Schematic diagrams showing the formation of banding a) and creation of large-scale chemical variation in the flow b). In a) high-density stippling denotes "raised" bands ~ 3 cm apart which are high in quenched glass whereas bands with low stippling form "eroded" bands in weathered outcrop and represent areas where un-quenched residual liquid has been, or is in the process of being, extracted. Thus the top of the column at time l shows moving magma eroding, and extracting interstitial liquid from fine-grained crystal mush. Like air passing over a plane's wing, magma movement causes a low-pressure zone immediately above the mush due to the Bernoulli effect, causing the extraction of interstitial liquid. When magma movement stops between pulses such as at time m (right-hand column), a new quench (raised) layer forms and the crystallization front moves further into the flow. Diagram b) shows the lower 1 to 2 metres of layer 2 forming, with low-density, residual liquid creating buoyant plumes that rise from the magma - lower crystal mush interface. In this diagram the amount of stippling reflects the degree of crystallinity (temperature below the liquidus) which increases (temperature decreases) away from the crystallization front.

due to early mixing in the lava lake (Helz *et al.* 1989). Possibly the flow at Henley Harbour was not thick enough to effect thorough mixing and homogenization. Increasing evidence suggests that large continental flood basalt flows form due to prolonged (months to years) eruption of magma causing flow inflation over time (Hon *et al.* 1994; Self *et al.* 1996). Obviously there is the potential for chemical zoning in a flow if magma composition changes. Certainly large variations in composition occur across individual Long Range Dykes that far exceed variation in the flow at Henley Harbour (Greenough and Owen 1992). Similar chemical variability in other continental dykes has been related to the tendency for increasingly evolved magma to move through dyke conduits over time (days to weeks; Greenough *et al.* 1989). Possibly the first magma extruded at Henley Harbour was more evolved than what later formed layer 2. This model also requires that magma compositions fortuitously returned to more evolved, layer 1-like compositions, to form layer 3.

Another possibility is that layer 1 reflects the initial magma composition whereas layer 2 is more mafic due to phenocryst accumulation. Obviously the chemical change between layers 1 and 2 is small but it is not clear that layer 2 is appreciably more phenocryst rich. As with the chemical variation between bands in layer 1, attempts to use observed phenocryst compositions to explain chemical differences between average layer 2 (parent) and an average of the raised bands from layer 1 were not satisfactory.

The best mass balance solutions required the most Fe-rich augite and Na-rich plagioclase phenocryst compositions measured, which seems unreasonable. Even then the lowest r^2 values were quite high (0.7) given how similar the two rocks are.

Possibly layer 2 experienced the removal of interstitial melt through the diapiric rise of low density Fe-depleted residual magma "blebs" formed by the crystallization of an Fe-rich phase (e.g., Fe-Ti oxides, olivine, or pyroxene) at the lower crystallization front. Such a model, using re-equilibrated olivine to remove Fe, elegantly accounts for incompatible element enrichments and Fe depletion high in Kilauea Iki Lava Lake and complimentary depletions/Fe enrichment lower in the lake (Helz *et al.* 1989). Alternatively, mass transfer through the formation of low-density vesicle cylinders, an apparently common process in basaltic lava flows (Helz *et al.* 1989; Greenough *et al.* 1999), could remove residual liquid from layer 2 and move it higher in the flow. Either of these models helps explain why both sets of mass balance calculations (banding and inter-layer calculations) require removal of Na-rich plagioclase and Mg-poor augite not seen in phenocrysts. The right combination of such phases would approximate the composition of a removed residual liquid.

As argued above, mass balance calculations indicate that compositional variations between bands in layer 1 and between large-scale layers in the flow cannot be explained by removal or accumulation of observed phenocryst compositions. If, however,

chemical variation is related to mass movement of residual liquids from a single parent magma, and the liquids were extracted with varying efficiency after the same percentage of crystallization, then mass balance calculations to ascertain the composition of these liquids, using different rocks from the flow, should yield similar evolved liquid compositions. To test this model, calculations used the average of the most evolved (lowest MgO) rock composition in the flow (average raised sample) and the two more “primitive” compositions (average eroded sample and average layer 2). Note that the “primitive” rocks represent samples that lost different amounts of a residual liquid. The compositions of the two theoretical evolved liquids were calculated according to the following equations:

$$\text{Average Layer 2} + \text{Liquid 1} = \text{Average Raised Sample}$$

$$\text{Average Eroded Sample} + \text{Liquid 2} = \text{Average Raised Sample}$$

Beginning with a crudely interpolated composition for each liquid, the mass balance calculations were performed repeatedly, each time modifying the liquid composition until the sum of squared residuals equalled zero. Calculation results together with the average starting rock compositions appear in Table 2. Despite using the two most extreme cases possible, the small chemical differences between the average eroded and average raised samples, and the large differences between the average raised sample and average layer 2, the calculated compositions of liquids 1 and 2 are remarkably similar (Table 2). Further, the liquids have the requisite characteristics of a residual liquid: higher SiO₂ and incompatible element (TiO₂, Na₂O, and K₂O) concentrations and lower MgO. A smaller amount of liquid is required to explain the differences between average eroded and average raised samples (36 g per 100 g of average eroded sample), than between the other pair of samples (80 g per 100 g of average layer 2; Table 2). The similarity of these calculated liquids may be fortuitous, but it suggests that movement of residual liquid on a cm- and m-scale is a viable method of explaining chemical variation within the flow. This model, summarized in Fig. 11, suggests that eroded bands of layer 1 appear more “primitive” than raised samples because evolved interstitial liquid was removed from the former. During the crystallization of layer 2, similar residual liquids formed buoyant plumes that rose from the magma-crystal mush interface, leaving layer 2 depleted in residual liquid (Fig. 11).

SUMMARY AND CONCLUSIONS

The lack of textural evidence for flow tops in the stratigraphic section examined, and similarity of all samples collected at Henley Harbour, indicates that large-scale layering does not reflect the existence of multiple flows. Chemical similarities to the Long Range Dykes confirm that the flow represents the surficial expression of dyke magmatism (Fig. 6). The chemical similarity among samples approaches analytical uncertainty and complicates interpretation of chemical data (Fig. 5, 7). Multidimensional scaling (MDS) improves effective precision by allowing simultaneous evaluation of the entire data set. It shows that, in general, ele-

Table 2: Calculated liquids and input data

	AVRAI	AVERO	AVLAY2	LIQ1	LIQ2
Wt. %					
SiO ₂	51.84	51.35	50.19	53.91	53.17
TiO ₂	1.35	1.34	1.29	1.43	1.40
Al ₂ O ₃	14.69	14.75	15.34	13.88	14.52
FeO	10.04	10.15	10.44	9.54	9.72
MnO	0.20	0.18	0.22	0.17	0.22
MgO	7.09	7.70	8.34	5.51	5.40
CaO	11.68	11.52	11.51	11.89	12.13
Na ₂ O	2.28	2.23	2.12	2.49	2.44
K ₂ O	0.69	0.64	0.43	1.02	0.86
P ₂ O ₅	0.13	0.13	0.12	0.15	0.15
Mass1	180	100			80
Mass2	136		100	36	

Notes: Oxide wt. % calculated to 100% volatile free. Total Fe as FeO. AVRAI, average raised sample; AVERO, average eroded sample; AVLAY2, average layer 2; LIQ1= Liquid 1 and LIQ2 = Liquid 2 as discussed in text. Mass1 and Mass2 give the relative masses of rocks and calculated liquids.

ment behaviour in the data set is consistent with igneous processes controlling overall chemical variability (Fig. 8). MDS also suggests that large scale layering apparent in outcrop has a chemical fingerprint (Fig. 9). Averages (Table 1) indicate that layer 1 (bottom) is less primitive than layer 2, and the single layer 3 sample lies compositionally in between, but closer to layer 1 (Fig. 9). Mass balance calculations show that chemical variability between layers cannot be explained by phenocryst accumulation or removal because unreasonable (more-evolved) mineral compositions are required to effect low r^2 values. Chemical variation may reflect changes in the composition of extruded magma during formation of an inflated flow. However, our preferred model, supported by mass balance calculations, attributes variability to the convective redistribution of evolved residual liquid formed from a single parental magma (Fig. 11). Analogies with other thick flows suggest that redistribution was effected by low density vesicle plumes, or Fe-depleted residual magma plumes, that rose from the lower, advancing crystallization front to higher levels in the flow.

MDS confirms that cm-scale banding expressed as 10's of m-long raised and eroded bands in outcrop also show a chemical signature (Fig. 10). As with the large scale layering, observed phenocryst compositions cannot account for the chemical variation between bands. However, mass balance calculations indicate that variations in composition between bands, like variations between large-scale layers in the flow, can be accounted for by movement of chemically evolved residual liquids that closely resembled one another in composition. Alignment of microlites in the bands and variations in the quantity of interstitial alteration minerals now replacing glass suggest that the cm-scale banding is related to magma movement during extrusion/emplacement of the flow (Fig. 11). Banding (layering) of this type may be common in basaltic lava flows, but field identification could require the combined subtle effects of glass alteration accentuated by weathering. The use of geochemistry to identify this type of layering is complicated

by the fact that bands are so chemically similar that it takes very powerful statistical methods, such as multidimensional scaling, to recognize it.

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